

Photon-induced desorption of large species at low temperature in TNOs

H. Carrascosa¹, G. A. Cruz-Díaz^{2,3}, G. M. Muñoz Caro¹, E. Dartois⁴, Y. -J. Chen⁵

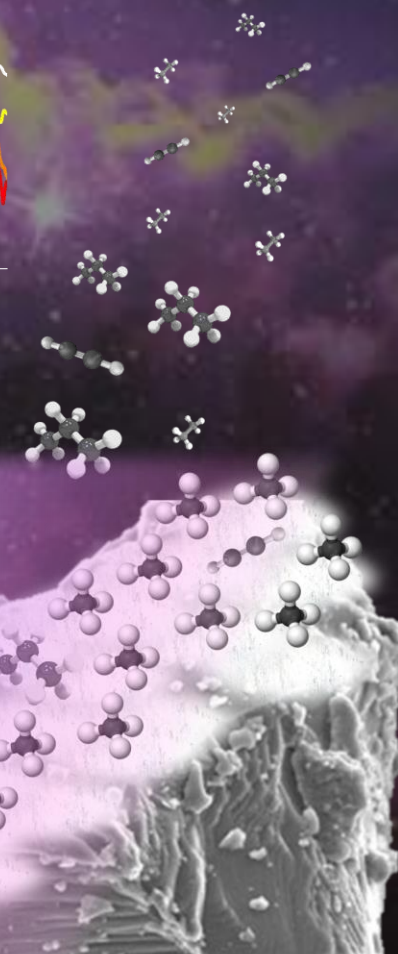
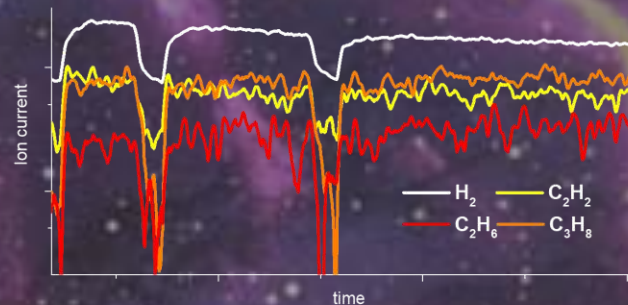
¹CAB, ²NASA Ames, ³Bay Area, ⁴ISMO, ⁵NCU MNRAS, 493, 821-829 (2020). DOI: [10.1093/mnras/staa334](https://doi.org/10.1093/mnras/staa334)

Some trans-Neptunian objects (TNOs) reveal the presence of abundant methane (CH₄) ice. Moreover, relatively large molecules, such as ethane (C₂H₆) or propane (C₃H₈), were detected in the atmosphere of TNOs, such as Pluto or Triton.

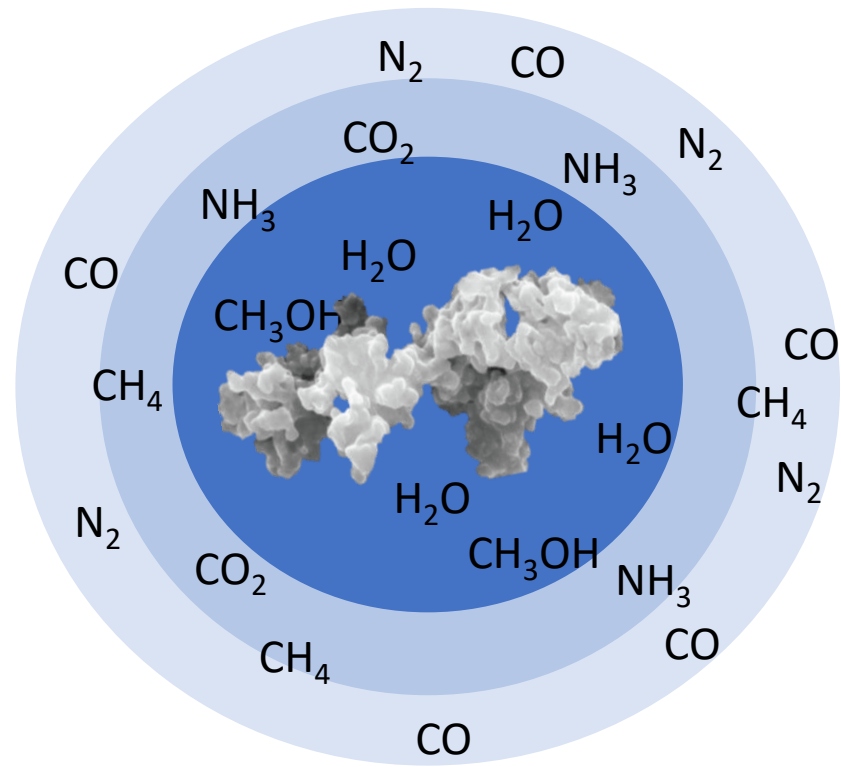
The present work aims to understand the formation and possible desorption mechanism which allows large molecules to be present in the gas phase at temperatures where their thermal desorption is negligible.

Experiments were carried out in ISAC, an ultra-high vacuum chamber equipped with a cryostat and an UV-lamp. Infrared spectroscopy and quadrupole mass spectrometry were used to monitor the solid and gas phase, respectively.

UV-irradiation of CH₄ ice led to formation of hydrocarbons. Photodesorption of C₂H₂ and photochemidesorption of C₂H₆ and C₃H₈ was observed, the latter implies an immediate ejection of these species after formation on the CH₄ ice surface.



Context: CH₄ ice mantles



Pre-cometary ice mantle

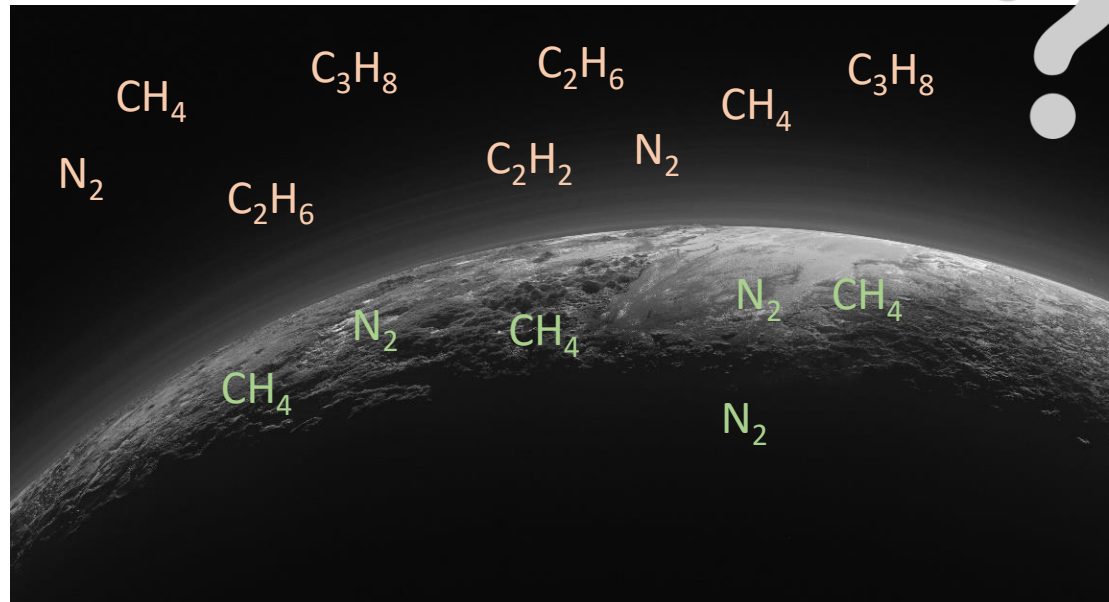
Interstellar medium



The low temperature in TNOs inhibits the presence of large molecules in the atmosphere. Methane is the only hydrocarbon able to thermally desorb to the gas phase.

How is it possible that ethane, propane, or acetylene have been detected in TNOs' atmospheres?

Transneptunian objects



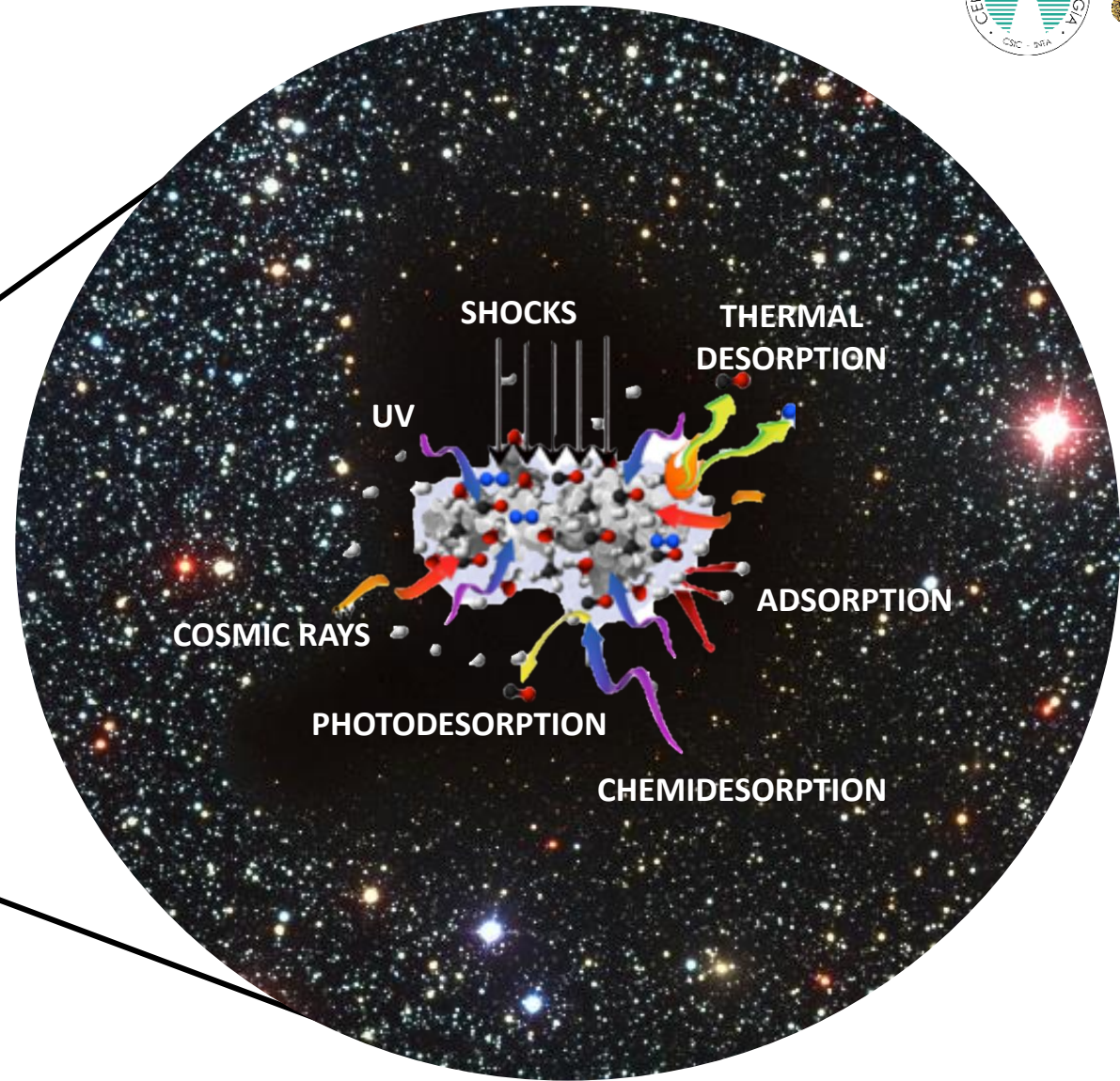
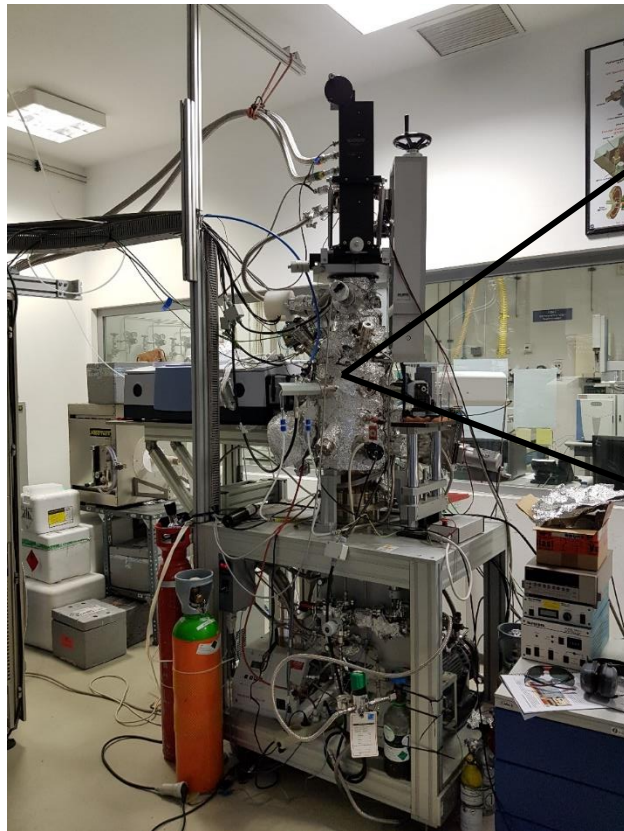
Methodology: ISAC

Reproduces the conditions in the ISM:

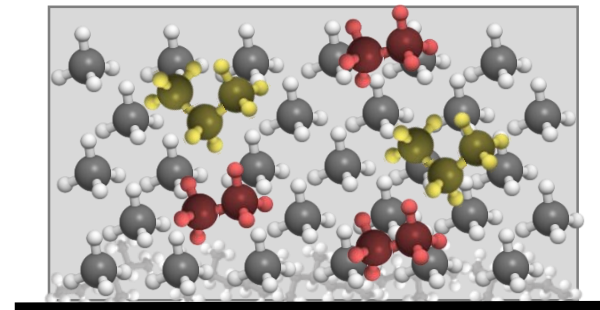
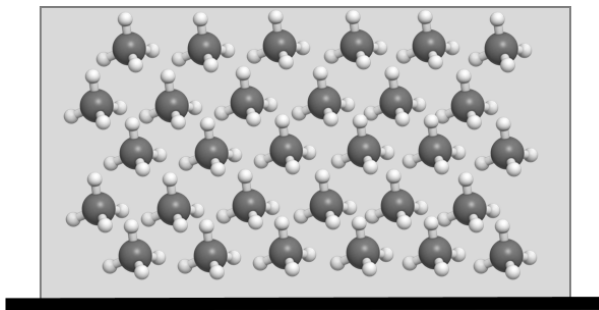
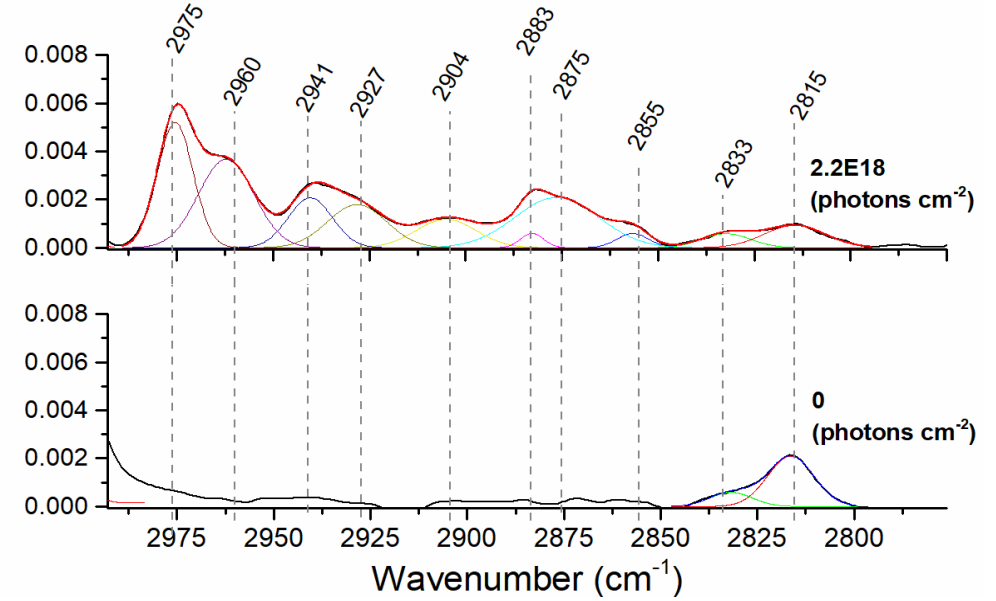
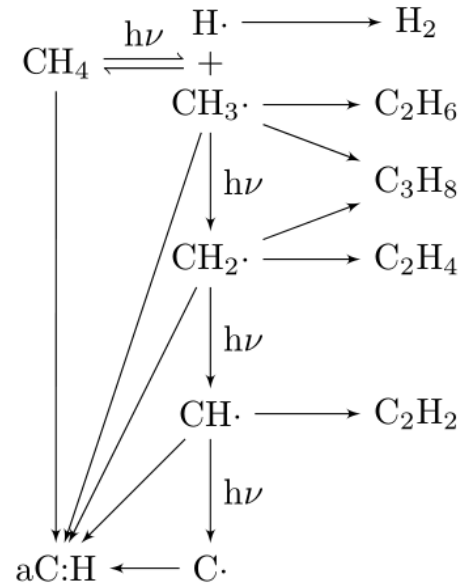
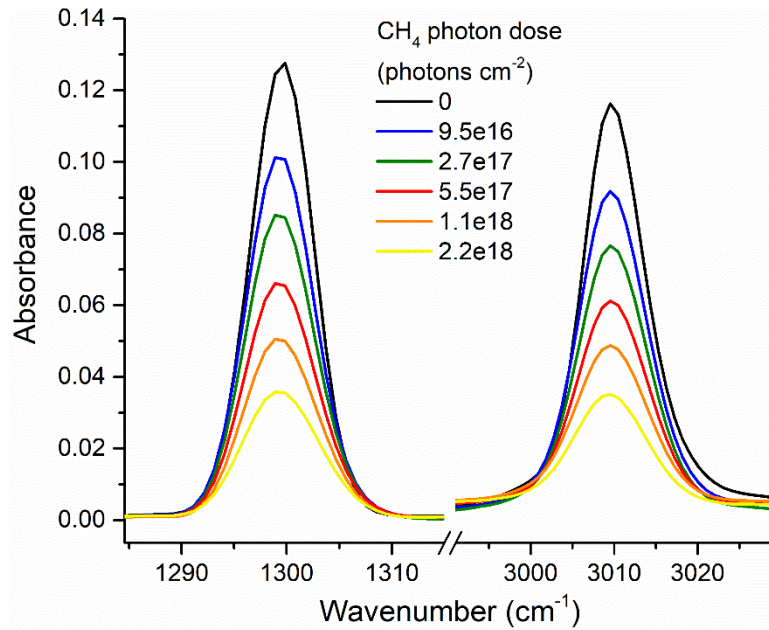
UHV $\rightarrow 10^{-11}$ mbar

T $\rightarrow 8$ K

UV \rightarrow MDHL

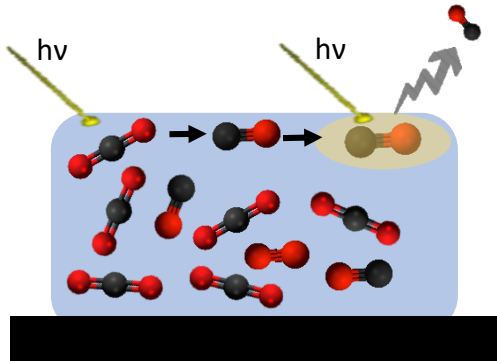


Results: formation of new species



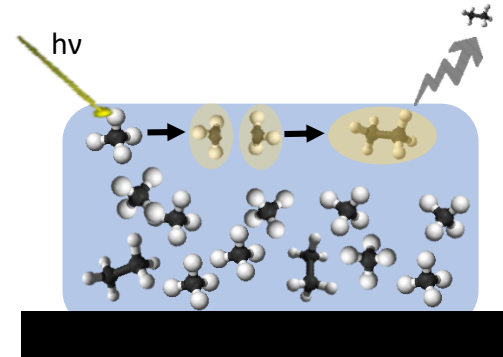
Results: photon-induced desorption mechanisms

Photodesorption



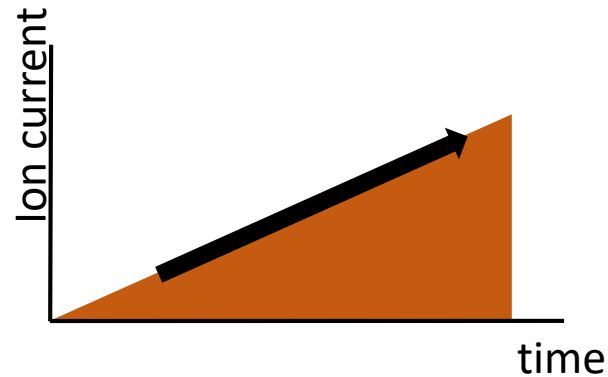
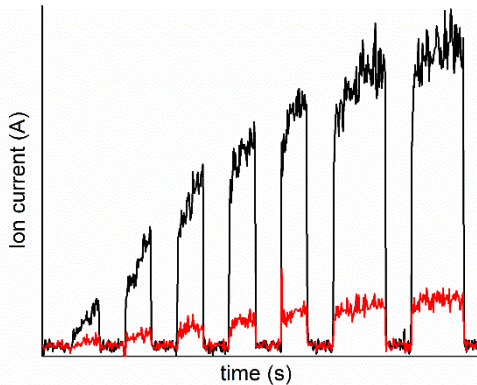
Molecules in the surface **receiving photon energy** to break intermolecular forces and desorb

Photochemical desorption

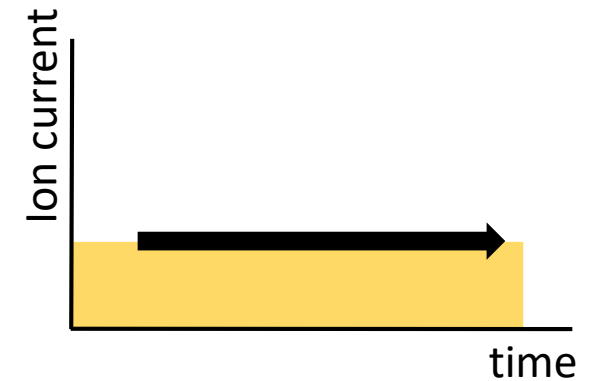
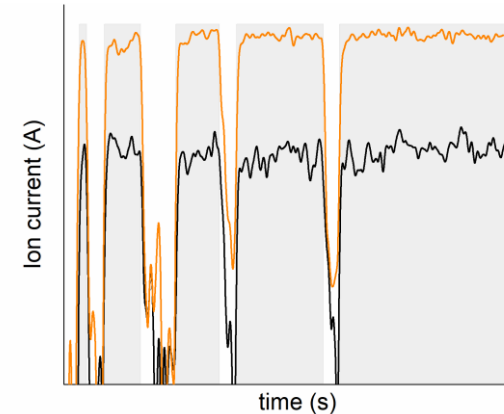


Molecules formed in the surface **releasing their formation energy** to break intermolecular forces and desorb

CO and O₂ from CO₂ ice



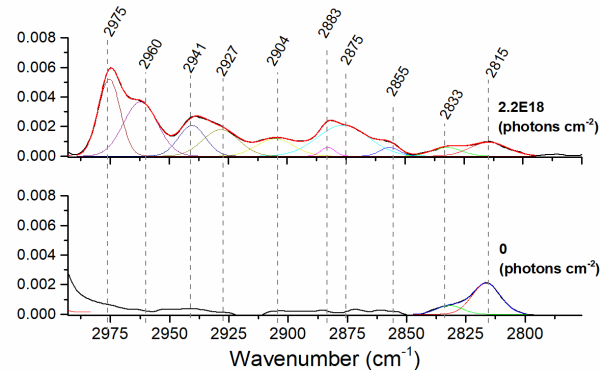
C₂H₆ and C₃H₈ from CH₄ ice



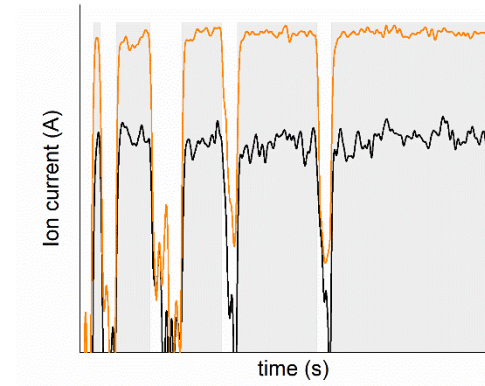
Impacts and prospects for the future

A better understanding of CH_4 ice processes under UV radiation is a first step to explore the chemical complexity in more realistic pre-cometary ice mantles containing different molecules.

Hydrocarbons synthesis

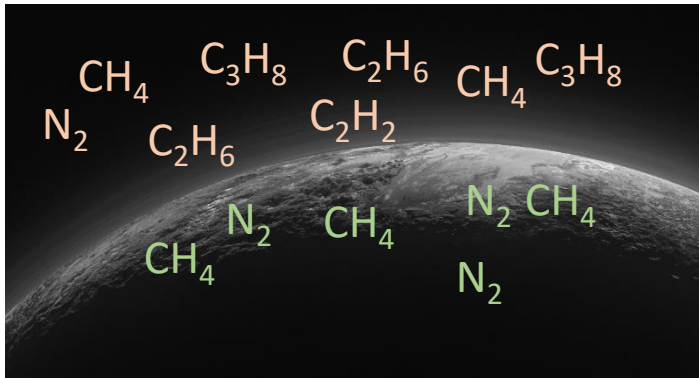


Photochemical desorption



Photochemical desorption can be applied to study the gas phase abundances of large molecules detected in different bodies.

Gas phase abundance in TNOs agrees well with the observations



Abundance in Pluto's atmosphere (Young et al. 2018)

$\text{C}_2\text{H}_6 > \text{C}_2\text{H}_2 > \text{C}_2\text{H}_4$ (C_3H_8 was not measured)

$\text{C}_2\text{H}_6 > \text{C}_3\text{H}_8 > \text{C}_2\text{H}_2 > \text{C}_2\text{H}_4$ (non-detected)

Photon-induced desorption in our experiments (Carrascosa et al. 2020)